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Evaluation of Epoxy Modification in Bitumen

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- 17 18

19 Abstract: In epoxy-bitumen systems, temperature contributes to the development of physiochemical and mechanical characteristics during curing (chemical hardening) and at the 20 long-term service (oxidative aging) of pavement structures. Hardening (i.e., chemical or 21 22 oxidative controlled) is a complex phenomenon in epoxy-modified binders and the in-depth exploration of incorporating epoxy chemistry in bitumen is needed to understand the 23 evolution of the properties of these binders in time. Within this framework, changes in two 24 25 different diluted epoxy-modified binders after oven-hardening for various time scales were analysed by means of Fourier transform infrared spectroscopy, modulated dynamic scanning 26 27 calorimetry and dynamic shear rheometry. The chemical, thermal and mechanical properties 28 of hardened binders were compared showing remarkable differences in their response under various conditions. The degree of aging was dependent on the level of epoxy modification in 29 bitumen. It was found that the sulfoxide compounds are the most representative index for 30 31 assessing the oxidative-controlled hardening of epoxy-modified bitumens. The phase angle gives useful information for assessing the long-term aging of modified binders demonstrating 32 a linear relationship between the phase angle and sulfoxide index changes with respect to 33 oxidative hardening. The improved performance (i.e., higher tensile strength, flexibility and 34 enhanced longevity) of newly modified binders was demonstrated as well. Overall, the 35 enhanced resistance against aging in combination with the superior mechanical characteristics 36 when the epoxy modification is implemented in bitumen promises a very effective 37 38 technology for developing long-lasting pavement materials.

39 INTRODUCTION

40 Studies of bitumen modification technologies have been undertaken in the past and 41 predominantly thermoplastic polymers (i.e., block polyolefins co-polymers) are nowadays 42 extensively used as modifiers to enhance the material durability both at low and high 43 temperatures (*1-4*). Nevertheless, the exploration of new modification approaches has been at 44 the forefront of pavement engineering research efforts due to the increasing demand on 45 developing pavement materials with improved long-lasting characteristics.

The use of epoxy modifiers has attracted the attention of the pavement scientists and 46 practitioners (5-10). Initially, the epoxy modifiers were utilized to minimize the permanent 47 deformation of bituminous materials because the thermoset nature of epoxies guarantees 48 excellent rut resistance (5). It was also suggested that use of polymeric structure of epoxy in 49 50 the bituminous matrix can enhance the resistance against aging of bitumen (10). However, despite experience of practitioners on implementing epoxy modifiers in bituminous materials, 51 a general lack of knowledge still exists about the impact of these modifiers on the 52 physiochemical characteristics and the long-term performance of bitumen. Properties, such as 53 tensile strength or glass transition temperature, have been evaluated in the past (11-18) but 54 55 with limited studies of chemistry-related hardening phenomena, such as curing (i.e., chemical-controlled hardening - CH) or aging (i.e., oxidative-controlled hardening -OH). 56

The addition of epoxy modifier in bituminous binders is a relatively new modification 57 58 technology and the crucial epoxy polymerization-induced changes of the epoxy-bituminous 59 materials are not fully understood. Different phenomena take place when different amounts of epoxy are incorporated into bitumen and they are dependent on the material hardening 60 61 conditions (20, 21). The concentration of epoxy resinous part in bitumen controls the microstructural characteristics of modified blends and might influence the longevity of these 62 materials. In this paper, the chemical compounds and their reaction products generated under 63 64 various hardening conditions were studied to reveal the time dependency of molecular microstructures of modified bitumen. Special attention was given to the evaluation of 65 physiochemical characteristics and the mechanical properties of epoxy bituminous binders. 66 Overall, the studies reported in this paper focused on establishing a testing framework 67 specially designed for epoxy-bitumens by linking the chemistry-related phenomena with their 68 mechanical performance. 69

70

71 MATERIALS AND PREPARATION

The epoxy-based modifier is a commercial product supplied by ChemCo Systems, USA and 72 named epoxy asphalt. This modifier is formulated from two liquid parts free from solvents; 73 74 (i) the part A (epoxy resin formed from epichlorhydrin and bisphenol-A) and (ii) part B 75 (blend of fatty acid hardening agent and 70 pen bitumen). When part A and B (both reacting systems) are mixed into bitumen as modifying elements, the dispersion of the modifier phase 76 77 inside the bitumen is achieved. In reality, the monomers of part B react with the epoxide 78 groups of part A to produce covalent bonds in bitumen and the molecular chains are 79 crosslinked through the polymerization process.

80 In this study, all the samples were prepared by mixing part A and B at weight ratio of 20:80, according to the supplier. Unless otherwise stated the two components, part A and B, 81 were oven-heated separately for 1 hour, to 85°C and 110°C, respectively. After that one hour, 82 they were mixed together for approximately 10 to 20 seconds and this epoxy modifier was 83 mixed further with an already pre-heated unaged 70-100 pengrade bitumen at 120°C, on the 84 basis of the supplier's recommendations. Two epoxy modification levels of bitumen were 85 86 studied; by adding 20 and 50 % wt. of modifier with replacing equivalent amount of bitumen. In other words, two epoxy-modified bituminous binders were formed with weight ratio of 87 20:80 (EB20) and 50:50 (EB50) of epoxy and bitumen, respectively. The study of these two 88

89 modification levels is linked also with an effort to generate economically feasible and long-90 lasting paving materials since the initial price of supplied system, named epoxy modifier, is 91 high. Finally, all the prepared samples were placed in a refrigerator at -10°C to prevent any 92 further reaction.

The hardening of the binders was simulated in a draft oven under atmospheric pressure 93 over various lengths of time. At a temperature of 130°C and at 1.0-kPa hardening times of 0, 94 95 2, 5, 8, 24, 120, 240 and 480 hrs, were compared. The same materials were aged in a pressure-aging vessel (PAV) on pans with 140-mm in diameter and 3-mm film thickness at 96 97 2.1-MPa pressure with oxygen (AASHTO MP1). PAV aging time for 100°C was 20 hrs. After each hardening period, the physiochemical and mechanical properties were measured as 98 function of time through Fourier Transform Infrared (FTIR) spectroscopy, modulated 99 100 dynamic scanning calorimetry (MDSC) and dynamic shear rheometer (DSR). The testing 101 methods used for this research are presented in the following sub-sections.

102

103 EXPERIMENTAL METHODS

104

105 Chemical and Physical Characteristics during Hardening

The chemical properties of the hardened epoxy modified bitumens (EBs) were determined 106 with a Perkin-Elmer Spectrum FT-IR spectrometer equipped with an Attenuated Total 107 Reflectance (ATR) fixture. The hardening of materials was analysed after the time intervals 108 used. The FTIR spectrometer was used to investigate the change of the molecular structure of 109 the binders after each hardening time and to provide information about CH and OH of EBs. 110 The FT-IR spectra with wavenumber from 4,000 to 600 cm⁻¹ were recorded and collected for 111 all the samples. A certain amount of material was placed directly on the ATR crystal pedestal 112 and pressed with a constant force to ensure proper contact to the surface. A minimum of three 113 sub-samples were investigated for each sample and 20 scans per sub-sample were performed 114 with a fixed instrument resolution of 4 cm⁻¹ (22, 23). Fig. 1(a) shows the FT-IR spectra of 115 base (EB0) and epoxy modified bitumen (EB50) recorded after 480 hrs at 130°C. 116

For OH of bitumen, the carbonyl (1753-1660 cm⁻¹) and sulfoxide index (1047-995 cm⁻¹) 117 are normally used to follow the aging in bitumen (24, 25). Their values were calculated by 118 using the area method representing the extent of age. The calculation is performed by 119 dividing the area under a specific location of the spectrum by the sum of other specific areas. 120 However, carbonyl groups (C=O) play an important role in the CH hardening of EBs as well. 121 During CH, oxirane groups (C2H4O) (at 917 cm⁻¹) react with the carbonyl acid group 122 (RCOOH) (at 1709 cm⁻¹) resulting into ester (RCOOR'') (at 1735 cm⁻¹) and ether (ROR') (at 123 1040 cm⁻¹). The potential contribution of methylene (H2C) (3050 cm⁻¹) is not considered in 124 these cases. Fundamental analysis was performed using the absorbance intensity method of 125 peak difference (Fig. 1(b)) after the CH and OH hardening times. 126

Besides the chemical changes in bitumens under various conditions, the materials also undergo microstructural transformations. A parameter that describes these temperaturerelated transformations is the glass transition temperature (T_g) . Bitumens show glassy behavior below T_g , the rubbery characteristics could be shown above T_g . Previous studies have shown a range of T_g values for epoxy-bitumens from -8 to 50°C (11-16), with neat binders to show a range between -40 to 0°C (19).

In this study, in addition to chemical evaluation of involving compounds through CH, also the phase transition temperature T_g of EBs was measured by MDSC under a nitrogen atmosphere with 40°C/min heating rate from -50 to 200°C. When the temperature of 200°C was reached the samples were kept at this temperature for 2 min. MDSC is a thermal analysis technique to determine the temperature and the heat flow associated with material transitions as function of temperature and time. After the preparation and hardening of samples, they were sealed in aluminum pans, and the CH behaviour was assessed by non-isothermal experiments. The peak and mid-point of the heat flow shift denotes the T_g of materials. At least two measurements were performed in each sample. Samples were hardening at different hardening times from 0 to 180 min of 60 min time step. The MDSC curves of the studied EBs during the heating processes are shown in **Fig. 2**, where T_g was characterised as the temperature at which the first slope alteration took place, which coincides to an endothermic peak in the first derivative plots.

146

147 Mechanical Characteristics during Hardening

To investigate the evolution of mechanical characteristics during curing (i.e., CH), the complex modulus at 10 Hz was measured for a time window between 0 to 180 min at the chosen hardening temperature of 130°C under isothermal conditions. DSR measurements were performed using the parallel plate test configuration with disposable plates of 25-mm diameter, and 1-mm sample thickness.

Additionally, viscoelastic characteristics, such as complex modulus and phase angle, were 153 used as reliable indicators for describing the aging behaviour (i.e., OH) of crosslinked epoxy 154 155 within the bitumen. Isothermal frequency sweep measurements were performed using a dynamic shear rheometer (DSR, Anton Paar, EC Twist 502) at different temperatures that 156 ranged from -10 to 60°C. The parallel plate testing geometry was used as well to evaluate the 157 viscoelastic properties of the different binders after different hardening periods. Plates of 8-158 mm diameter with a 2-m gap were used at temperatures below of 20°C, while at temperature 159 above 30°C plates of 25-mm diameter with a 1-mm sample gap were used. The properties 160 161 were measured at frequencies of 0.1-10 Hz at temperature steps of 10°C from -10 to 60°C. Sample was placed onto the bottom plate at the desired test temperature ($\pm 0.1^{\circ}$ C). The 162 samples of base binder were treated in the same way and master curves were constructed. For 163 the master curves, the time-temperature superposition model was used to shift all 164 temperatures to a reference temperature of 30°C. 165

166

167 Performance under Monotonic and Cyclic Loading

The tensile strength has been investigated in the past showing various strength values from 168 1.25 to 9.90 MPa at standard test conditions (11, 13-18). In this paper, after curing, the 169 strength of these binders was determined under controlled isothermal temperature conditions 170 at 0°C. The samples were prepared, poured into a Teflon mould, and taken out of the storage 171 container any excess material was trimmed of and fixed with clamps in the DSR. The 172 cylindrical sample (height 10-mm, diameter 6-mm) was loaded in tension with a 173 174 displacement speed of 0.05 mm/s. The tensile strength was calculated on the base of the sample geometric configuration and the calculated value can be used to compare the results 175 176 of different modifications with bitumen.

177 Finally, the DSR was used to assess the long-term performance of epoxy binders via a shear fatigue test, since the ultimate scope of incorporating this type of modifier is to increase 178 the longevity of bituminous materials. The DSR testing geometry of 8-mm diameter was used 179 with 2-mm gap and time sweep tests at a constant temperature of 0°C and a frequency of 10 180 Hz were performed at three different shear stress levels (i.e., 0.4, 0.5 and 0.6 MPa). At each 181 stress level, three fatigue tests were done for each modification. A decrease of the shear 182 183 modulus to 50% of its initial value was used as criteria to determine the load repetitions till failure. 184

185 **RESULTS AND DISCUSSION**

186

187 Oxidative controlled Hardening

In general, the properties of EBs are linked to specific changes in their chemical structure.
Initially, the transition borders from CH to OH phenomena, and the initiation of aging, were
identified by measuring the carbonyl and sulfoxide components.

Fig. 3(a) shows the carbonyl content calculated with area method as a function of oven-191 hardening time at 130°C for the studied binders. Along with the carbonyl content, the change 192 of sulfoxide due to age hardening was evaluated with the same method as well, and the 193 results depicting the relationship of sulfoxide over different times for EB0, EB20 and EB50 194 are shown in Fig. 3(b). At high time lengths, both sulfoxide and carbonyls can indicate OH. 195 These compounds increase as hardening time increases and the results gave a similar 196 tendency for all samples with a lower hardening effect for the modified binders. Therefore, it 197 seems that the modified binders are more resistant against aging with EB50 showing the 198 lowest OH sensitivity, see Fig. 3(b). EB50 shows a lower sulfoxide index than the base 199 bitumen (EB0). The carbonyl and sulfoxide compounds of EB0 increase more rapidly than 200 201 EB20 and EB50, and as consequence, higher epoxy modification levels provide lower aging indices. In addition to oven-hardening simulations, the long-term performance of EBs after the 202 PAV test seems promising as well when the sulfoxide index is used. According to the PAV 203 204 results, despite the inconsistencies of carbonyls generated from oven and PAV hardening (Fig. 3(a)), the sulfoxide compounds are equivalent to oven hardening at 130°C for 480 hrs (Fig. 205 **3(b)**). The patterns generated from the two methods demonstrate that the sulfoxide index gives 206 207 an effective way to describe the OH in EBs.

The viscoelasticity behaviour is determined by exploring the frequency-dependent material properties (i.e., complex shear modulus and phase angle) of the binders. Due to the fact that EBs consist of a wide range of temperature-dependent molecular structures with various polarities, the viscoelasticity of these binders is related completely on the interactions between the different structures and their chemistry.

Fig. 4 shows the complex modulus master curves for the binders after oven hardening at 213 130°C. These curves demonstrated the influence of OH on the viscoelastic behaviour of the 214 different EBs. It can be observed that the complex modulus increases slowly over the first 24 215 hrs of hardening, in which the network formation has been formed and the material hardens 216 due to curing, and then increases substantially as time increases up to 480 hrs. At relatively 217 low frequencies, the complex modulus increases continuously at all time lengths. The 218 modulus tends to merge together to the same value at higher frequencies for both EB20 and 219 220 EB50. The performance is clearly dominated by the inclusion of epoxy modifier into the bitumen. The slight difference between the modulus at EB20 and EB50 is due to that fact that 221 epoxy molecules need more time to link with each other at low modification levels. 222

223 The phase angle master curves are not as straight forward as the complex modulus mastercurves and the epoxy modifications in the bitumen showed considerable reductions in 224 phase angle. The phase angle is measured of the time lag between an applied stress and the 225 resulting strain and indicates whether a binder will behave as elastic-, visco-elastic or 226 viscous-dominated material. When the phase angle is low, the material is more elastic. For 227 EB20 and EB50, it can be seen that the modulus is higher at longer hardening times but the 228 slope of the phase angle of EB50 is different than of EB20. From the results it can be seen 229 also that the modifier had insignificant effects on the phase angle at relatively low 230 frequencies. The presence of phase angle plateaus at intermediate frequencies for EBs 231 232 indicates the epoxy molecular networks in binders. The modifier enhanced the elasticity of bituminous material. The material behaves more glassy when the hardening process has been 233

completed, due to the dominance of modifier in EB50. Measurements of phase angle aremore sensitive to chemical changes and thus the modification of bitumen than modulus.

The above findings have suggested that it could be interesting to link the evolution of 236 viscoelastic properties to the chemical characteristics of the binders during OH. Fig. 5(a) and 237 Fig. 5(b) show the relationship between the sulfoxide index and the changes in complex 238 modulus and phase angle at 10 Hz, respectively, indicating a linear relationship for all binders 239 over the OH extent. This performance indicates that the sulfoxide formation tends to correlate 240 with increased modulus and reduced phase angle as the binders harden. In addition to this 241 observation, the viscoelastic response of binders subjected to PAV hardening provided almost 242 identical hardening with the oven hardening for 480hrs, with the EB50 and EB0 showing the 243 highest modulus and lowest phase angle, respectively. Finally, since the level of OH is of 244 primary concern when comparing hardened binders, it is reasonable to conclude that addition 245 of epoxy modifier generates more resistant binders against oxidative aging. However, the 246 lower the phase angle at the same stiffness, probably the more susceptible a binder becomes 247 to monotonic and cyclic fracture. For this reason, the tensile strength characteristics and the 248 fatigue life of studied binders were assessed as well, and the results are given in a later sub-249 250 section.

252 Chemical controlled Hardening

251

According to the previous observation (see **Fig. 2(a)**), the initial fluctuation of carbonyl indices at the beginning of hardening (from 0 to 8 hrs) may be related with the possible contribution of these compounds to CH. For this reason, the time between 0 to 8 hrs is assumed to be the time window for CH.

Fig. 6(a) demonstrates that the intensity of oxirane reduces with prolonged hardening time 257 for both EBs tending to converge as time increases. The epoxide rings (C2H4O) are opening 258 through the consumption of oxirane and new compounds are formed. The hardener is 259 considered to be an acid-type since the carbonyl acid group decreased over time similarly 260 with the oxirane group. This performance indicates that the carbonyl acid and oxirane 261 precursors (the reactive groups) were consumed almost completely after approximately 5 hrs 262 at 130°C. Ester (RCOOR') and ether (ROR') compounds increase initially while remaining 263 constant after a time period indicating that esterification and etherification reactions take 264 place during the CH. Therefore, it is assumed that oxirane and carbonyl acid groups reacted 265 together to create the crosslinked network into the bituminous phase and formed the ester and 266 ether groups (Fig. 6(b)). Moreover, the level of epoxy modification affects the CH 267 accelerating the reactions when higher epoxy amounts were used. The network of epoxy 268 269 modifier (as resulted from the polymerization of epoxy resin in part A of modifier with the assistance of hydroxyl functionalities of hardener in part B) is formed slightly faster when 270 higher modification levels are applied, mainly because there is less bitumen available to 271 272 prevent the network evolution. Moreover, it can be seen that longer hardening times produces a higher curing increment. This can be seen from the steeper curve shape at the higher 273 modification level (EB50) than at a lower (EB20). 274

275 FTIR spectrometer, MDSC and DSR analyses were conducted to explore the increase of T_g and modulus of crosslinking epoxy binders within the CH window. As shown in Fig. 7, 276 the presence of the epoxy modifier in the bitumen causes an increase of modulus and T_g 277 whereas longer hardening times also increase both parameters. On the one hand, the increase 278 of mechanical and physical parameters as depicted in Fig. 7 is related to the increase of 279 crosslinking density until a point that represents the ultimate value corresponding to the 280 281 completion of CH. Also, these results show that both material parameters are very sensitive to small changes in the microstructure of bitumen at a constant (high) temperature, and 282 consequently of the crosslinking network of the reacting modifier. 283

On the other hand and from the molecular viewpoint, bitumen consists of a plethora of 284 molecules classified into asphaltene and maltene fractions. Asphaltenes (polar compounds) 285 are dissolved into the maltenes medium of various in terms of polarity compounds. In 286 addition to the fact that at very low temperatures, the T_g of bitumen correlates mainly to the 287 non-polar compounds of maltene medium (T_g) (26), the polar epoxy chains-asphaltene 288 interactions could result increase of mobility of non-polar moieties leading to T_g decrease. 289 However, from Fig. 7, the T_g increases over time during CH showing that the bitumen almost 290 does not participate and the epoxy-asphaltene interaction is limited in CH. It is assumed that 291 292 bitumen reduces the contact opportunities between unreacted parts of modifier still available 293 in the matrix and hence EBs release less heat. In other words, bitumen inhibited the polymerization reactions between the two epoxy parts leading to reduction of maltenes 294 mobility and subsequently to T_g increase. In general, differences in polymer crosslinking 295 296 structures in fully hardened modified binders are related also to the polarization and orientation of reacted molecules and the reaction pathways. 297

298

299 Mechanical Performance of CH Hardened Binders

300 The tensile characteristics of fully chemically hardened samples (i.e., at 130°C after 5 hrs) under monotonic direct tensile load was examined as well. In general, the tensile strength of 301 the samples depends on the extent of the hardening reaction, the ratio of the unlinked 302 molecules to crosslinked molecules and the internal chemical structures. The modified 303 binders experienced ductile behavior, and thus exhibited higher flexibility characteristics than 304 305 the base bitumen (Fig. 8(a)). Modifier enhances greater toughness in the bituminous system 306 forming materials of high tensile failure resistance (i.e., high tensile strength), and this strength increased with the content of modifier into the bitumen (Fig. 8(b)). 307

Finally, the fully hardened samples (i.e., at 130°C after 5 hrs) were exposed to cyclic loading at three different shear stress levels in the DSR to assess the fatigue properties of these newly developed binders. As shown in **Fig. 9**, the fatigue life at all three stress levels is higher for the epoxy modified binders than for the unmodified bitumen (EB0). The difference at lower stress levels is huge. In conclusion, not only the toughness of the modified bituminous also improved the fatigue performance. Combination of these properties with low temperature behaviour can strongly increase cracking resistance at all temperatures.

315

316 SUMMARY AND CONCLUSIONS

In this paper, epoxy polymerization-induced physiochemical phenomena have been related to 317 the mechanical properties, and two corollaries to the chemical analysis were made helping to 318 319 understand the behaviour of EBs. Firstly, the increase and the decrease of certain carbonyl compounds (increase of carbonyl ether and ester, decrease of carbonyl acid) at the beginning 320 of hardening (i.e., the first 5 hrs) indicate that the esterification and etherification (CH) are 321 taking place due to the reaction of hardener (part B of modifier) with the epoxy resin (part A 322 of modifier). Second, the inconsistency of carbonyls for the short and long time intervals in 323 the oven- and PAV-hardening leads to the conclusion of termination of chemical and the 324 initiation of age hardening after 5 hrs, respectively. The sulfoxide index is proposed to be an 325 efficient indicator for evaluating the aging extent of epoxy modified binders. With regards the 326 mechanical performance of EBs, the complex modulus of EBs is similar to base bitumen, and 327 328 EBs provide higher tensile strength, flexibility and enhanced fatigue performance.

328 EBs provide highe 329

330 ACKNOWLEDGEMENTS

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- 335

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408 LIST OF TABLES AND FIGURES

FIGURE 1 FTIR spectra of (a) base and epoxy modified bitumen (EB50), and (b) band
area and peak values and limits for carbonyl groups of EB50 at an early and a late
hardening stage

- 414 FIGURE 2 MDSC scans at 40°C/min from -50 to 200°C of base and epoxy modified 415 bitumen (EB50)
- 416
 417 FIGURE 3 Relationship of (a) carbonyl (C=0) and (b) sulfoxide (S-O) and time in oven
 418 at 130°C
- 419
 420 FIGURE 4 Master curves of complex modulus and phase angle of studied binders after
 421 OH for different times; (a) EB0, (b) EB20, and (c) EB50
- 422
 423 FIGURE 5 Relationship between sulfoxide compounds and viscoelastic properties over
 424 OH hardening: (a) complex modulus and (b) phase angle
- 426 FIGURE 6 Relationship of (a) reactants, and (b) products during oven-hardening in 427 time at 130 °C
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- FIGURE 9 Cyclic shear test results of fully chemically hardened binders at
 temperature of 0°C and frequency of 10 Hz
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FIGURE 1 FTIR spectra of (a) base and epoxy modified bitumen (EB50) at late
hardening stage, and (b) band area and peak values and limits for carbonyl groups of
EB50 at an early and a late hardening stage



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451 FIGURE 2 MDSC scans at 40°C/min from -50 to 200°C of base and epoxy modified 452 bitumen (EB50)



459 FIGURE 3 Relationship of (a) carbonyl (C=0) and (b) sulfoxide (S-O) and time in oven
460 at 130°C
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470 FIGURE 4 Master curves of complex modulus and phase angle of studied binders after

471 OH for different times; (a) EB0, (b) EB20, and (c) EB50



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